

Supplementary Material

1 Supplementary Data

1.1 Quantum Yield Measurement

The fluorescence quantum yields of the fluorophores were measured with a similar method to previously reported (Yang et al., 2017; Yang et al., 2018; Ma et al., 2020). The fluorescence spectra in the region of 900-1500 nm were measured by a spectrometer with a thermoelectrically cooled InGaAs detector (HORIBA Ihr320) under an 808 nm diode laser excitation (RMPC lasers, 180 mW). During emission measurements, one 850-nm short pass filter (Thorlabs) was used as the emission filter. The obtained emission spectra were further corrected by the detector sensitivity profile and the absorbance features of the filter. The fluorescence quantum yield was determined against the reference fluorophore **IR-FE** with a known quantum yield of 3.1% (Φ_{st}) in toluene, which was previously determined with **IR-26** of 0.050% as reference in dichloroethane. All samples were measured at 25 °C with optical density (OD) below 0.1 at 808 nm. The intensity read out from the InGaAs camera was a spectrally integrated total emission intensity in the 900-1400 nm region. Using the measured optical density (OD) at 808 nm and spectrally integrated fluorescence intensity (F), the quantum yield of the test sample can be calculated according to the following equation:

$$\Phi_x(\lambda) = \Phi_{st}(\lambda) \times \frac{F_x}{F_{st}} \times \frac{A_{st}(\lambda)}{A_x(\lambda)} \times \frac{\eta_x^2}{\eta_{st}^2} = \Phi_{st}(\lambda) \times \frac{F_x}{F_{st}} \times \frac{1 - 10^{-OD_{st}(\lambda)}}{1 - 10^{-OD_x(\lambda)}} \times \frac{\eta_x^2}{\eta_{st}^2}$$

Φ_{st} and F_{st} are data of the **IR-FE** standard, Φ_x and F_x are data of the studied sample. η is the refractive index of solvent.

1.2 Density Functional Theory Calculations

To reduce the computational cost, side chains on the benzene units are replaced by methoxy groups. The ground-state (S_0) geometries of structure-simplified **BGM6**, **BGP6** and **BGO6** were firstly optimized at the B3LYP/6-31G(d) level (Lee et al., 1988; Heyd et al., 2003) and then re-optimized at the tuned- ω B97XD*/6-31G(d) level. The corresponding range separation parameter (ω , in Bohr⁻¹) for each molecule was optimally tuned and listed in **Supplementary Table 1**. The excited-state (S_1) geometries of these molecules were optimized using the time dependent (TD)-tuned- ω B97XD*/6-31G(d) method (Runge and Gross, 1984). The HOMO and LUMO orbitals, absorption excitation

energies of these molecules were obtained at the TD-tuned- ω B97XD*/6-31G(d) level based on their optimized S_0 geometries. The emission excitation energies of these molecules were calculated at the TD-tuned- ω B97XD*/6-31G(d) level based on their optimized S_1 geometries. The polarizable continuum model (PCM) (Tomasi et al., 2005) was employed to take into account the effects of the solvents. All the calculations were performed using the Gaussian 16 software.

1.3 Molecular Dynamic Simulations

For each molecule, the structure was optimized at the PCM(water)-B3LYP/6-311G** level (Lee et al., 1988), and then restrained electrostatic potential (RESP) charges (Bayly et al., 1993) and the General Amber Force Field (GAFF) (Wang et al., 2004) were assigned for the optimized structure. To remove bad contacts before the simulation, 2000 steps of steepest descent followed by 8000 steps of conjugate gradient energy minimizations were carried out. All bonds with hydrogen atoms were fixed using the SHAKE algorithm (Ryckaert et al., 1977). The particle mesh Ewald method with an 8 Å cutoff in real space was used to calculate electrostatic interaction. A Langevin thermostat with a collision frequency of 1.0 ps^{-1} was used to regulate temperature. Isotropic pressure coupling with a relaxation time of 2 ps was used to maintain the pressure to 1 atm. All the MD simulations were performed by AMBER 18 program. The water molecules around the BBTD acceptor center in the effective contact distance ($R = 6 \text{ Å}$) are displayed as the explicit water model. The initial structure of fluorophore was immersed in the center of a truncated octahedral box of TIP3P (Jorgensen et al., 1983) water molecules, and all of the PEG atoms were no less than 8 Å from the boundary of the water box. The relaxed structure was then gently heated from 0 to 300 K in 50 ps and equilibrated for 50 ps with weak restraints on each molecule, which was equilibrated for another 500 ps at constant pressure without restraint. Production simulations were extended to 140 ns for each molecule and trajectories were saved every 2 ps.

1.4 Centrifugal Filtration of Fluorophores

Considering the large aggregation of as-prepared fluorophores can be removed using centrifugal filter, fluorophores through filtration of 30, 50 and 100 kDa molecular weight cut-off (MWCO) are intravenously injected to investigate the effect of fluorophores aggregation on excretion behavior. The pass percentages of fluorophores through different filters are estimated with the optical density (OD) values at the peak and summarized in **Supplementary Table 3**. Selected time points from video-rate NIR-II imaging of mice in the supine position after tail vein injection of as-prepared, 30 kDa and 100

2 Synthetic Procedures and Characterization Data for the Molecular Fluorophores

M1: R1 = R4 = -OC₆H₁₂Br, R2 = R3 = H
M2: R2 = R4 = -OC₆H₁₂Br, R1 = R3 = H
M3: R2 = R3 = -OC₆H₁₂Br, R1 = R4 = H
M4: R1 = R4 = -OC₆H₁₂Br, R2 = R3 = H
M5: R2 = R4 = -OC₆H₁₂Br, R1 = R3 = H
M6: R2 = R3 = -OC₆H₁₂Br, R1 = R4 = H
BGM6: R1 = R4 = -OC₆H₁₂Br, R2 = R3 = H
BGP6: R2 = R4 = -OC₆H₁₂Br, R1 = R3 = H
BGO6: R2 = R3 = -OC₆H₁₂Br, R1 = R4 = H

BGM6P: R2 = R3 = H, R1 = R4 =
BGP6P: R1 = R3 = H, R2 = R4 =
BGO6P: R1 = R4 = H, R2 = R3 =

Supplementary Scheme 1. Synthetic routes of **BGM6**, **BGP6** and **BGO6** and their PEGylated compounds **BGM6P**, **BGP6P** and **BGO6P**.

The synthetic procedures of **M1-M3** were similar to the reported procedures in our previous work (Wan et al., 2018).

Compound **M1** (yield 53%). ^1H NMR (500 MHz, CDCl_3), δ (ppm) 7.19 (t, $J = 6.5$ Hz, 1H), 7.15 (d, $J = 1.5$ Hz, 1H), 6.58 (d, $J = 1.5$ Hz, 2H), 6.30 (d, $J = 1.5$ Hz, 1H), 4.16 (m, 2H), 3.99 (m, 4H), 3.88 (m, 2H), 3.75 (m, 2H), 3.76 (m, 2H), 3.70 (m, 2H), 3.68 (m, 2H), 3.57 (m, 4H), 3.40 (m, 3H), 1.89 (m, 4H), 1.83 (m, 4H), 1.55 (m, 8H). ^{13}C NMR (126 MHz, CDCl_3), δ (ppm) 157.35, 156.11, 133.25, 128.65, 121.22, 113.07, 105.46, 97.85, 72.08, 70.19, 70.81, 70.72, 69.94, 69.30, 68.98, 59.18, 33.99, 32.80, 29.07, 27.94, 25.46. HRMS (ESI) calcd for $\text{C}_{29}\text{H}_{45}\text{O}_6\text{Br}^{\text{SI}}\text{BrS}$, ($[\text{M}+\text{H}]^+$) 681.1278, Found 681.1273.

Compound **M2** (yield 51%). ^1H NMR (500 MHz, CDCl_3), δ (ppm) 7.19 (s, 1H), 7.14 (d, $J = 1.5$ Hz, 1H), 6.87 (d, $J = 1.5$ Hz, 1H), 6.77 (s, 1H), 6.27 (d, $J = 1.5$ Hz, 1H), 4.15 (m, 2H), 3.99 (m, 4H), 3.95 (m, 2H), 3.92 (m, 2H), 3.87 (m, 2H), 3.76 (m, 2H), 3.70 (m, 2H), 3.68 (m, 2H), 3.57 (m, 2H), 3.45 (m, 3H), 3.40 (m, 2H), 1.91 (m, 6H), 1.80 (m, 2H), 1.55 (m, 8H). ^{13}C NMR (126 MHz, CDCl_3), δ (ppm) 156.81, 153.15, 149.72, 138.12, 123.98, 117.12, 114.47, 114.11, 114.08, 98.32, 72.07, 70.93, 70.80, 70.72, 69.86, 69.47, 69.40, 68.53, 59.20, 34.01, 33.99, 32.83, 32.78, 31.09, 29.32, 28.08, 28.02, 25.56, 25.44. HRMS (ESI) calcd for $\text{C}_{29}\text{H}_{45}\text{O}_6\text{Br}^{\text{SI}}\text{BrS}$, ($[\text{M}+\text{H}]^+$) 681.1278, Found 681.1273.

Compound **M3** (yield 64%). ^1H NMR (500 MHz, CDCl_3), δ (ppm) 7.04 (d, $J = 1.5$ Hz, 1H), 6.89 (d, $J = 1.5$ Hz, 1H), 6.85 (d, $J = 1.5$ Hz, 1H), 6.83 (s, 1H), 6.15 (d, $J = 1.5$ Hz, 1H), 4.14 (m, 2H), 4.04 (m, 2H), 3.98 (m, 2H), 3.85 (m, 2H), 3.83 (m, 2H), 3.74 (m, 4H), 3.70 (m, 2H), 3.66 (m, 2H), 3.64 (m, 4H), 3.40 (m, 2H), 1.91-1.81 (m, 8H), 1.54 (m, 8H). ^{13}C NMR (126 MHz, CDCl_3), δ (ppm) 157.45, 149.26, 149.14, 142.93, 127.80, 118.43, 115.02, 114.07, 111.61, 96.35, 72.04, 70.91, 70.77, 70.68, 69.79, 69.44, 69.20, 69.16, 59.14, 33.87, 32.84, 32.82, 29.22, 29.20, 28.04, 28.02, 25.38, 25.37. HRMS (ESI) calcd for $\text{C}_{29}\text{H}_{45}\text{O}_6\text{Br}^{\text{SI}}\text{BrS}$, ($[\text{M}+\text{H}]^+$) 681.1278, Found 681.1271.

The intermediates **M4-M6** were prepared with the similar procedures as follow.

General procedures for **M4-M6**. To a solution of compound **M1-M3** (680 mg, 1.0 mmol) in 60 mL THF at -78 °C under protection gas atmosphere, *n*-BuLi (1.6 M in hexane, 0.69 mL, 1.1 mmol) was added dropwise. After stirring at this temperature for another 1.5 h, tributyltin chloride (340 mg, 1.2

mmol) was added to the solution. Then the reaction was slowly warmed to room temperature and stirred overnight. After that, the mixture was poured into water and extracted twice with ethyl acetate, the combined organic phase was dried with MgSO_4 and evaporated in vacuo without further purification.

Product **BGM6**, **BGP6** and **BGO6** were prepared with the similar procedures.

General procedures for **BGM6**, **BGP6** and **BGO6**. To a solution of the crude compound **M4-M6** (1.1g, about 1.12 mmol) and BBT-Br (157 mg, 0.45 mmol) in toluene (15 mL) under protection gas atmosphere, $\text{Pd}(\text{PPh}_3)_2\text{Cl}_2$ (140 mg) was added. The mixture was stirred at 120 °C for 12 h. After cooling to room temperature, the mixture was poured into water and extracted twice with ethyl acetate. The organic phase was dried with MgSO_4 and evaporated in vacuo. The crude material was purified by silica gel column chromatography (PE/DCM = 4:1) to afford compound **BGM6**, **BGP6** and **BGO6** as dark green solid (yield ~45%).

Compound **BGM6** (yield 43%). ^1H NMR (500 MHz, CDCl_3), δ (ppm) 7.80 (s, 2H), 7.26-7.18 (m, 2H), 6.66 (d, $J = 1.5$ Hz, 2H), 4.36-4.33 (t, $J = 6.5$ Hz, 4H), 4.10-4.07 (m, 8H), 3.71-3.68 (m, 4H), 3.57-3.53 (m, 12H), 3.49-3.47 (m, 4H), 3.36-3.32 (m, 14H), 1.93-1.80 (m, 16H), 1.57-1.49 (m, 16H). ^{13}C NMR (126 MHz, CDCl_3), δ (ppm) 157.29, 154.87, 153.08, 137.02, 128.69, 119.63, 114.75, 114.12, 112.83, 105.45, 72.00, 70.88, 70.79, 70.65, 70.61, 70.23, 69.19, 59.12, 33.96, 32.80, 29.22, 28.07, 25.66. HRMS (ESI) calcd for $\text{C}_{64}\text{H}_{87}\text{O}_{12}\text{N}_4\text{Br}^{81}\text{BrS}_4$, ($[\text{M}+\text{H}]^+$) 1551.1890, Found 1551.1881.

Compound **BGP6** (yield 47%). ^1H NMR (500 MHz, CDCl_3), δ (ppm) 7.57 (s, 2H), 7.31 (s, 1H), 6.93-6.90 (d, $J = 1.5$ Hz, 1H), 6.84-6.81 (m, 1H), 4.40-4.38 (t, $J = 6.5$ Hz, 2H), 4.10-4.07 (m, 2H), 4.01-3.99 (m, 2H), 3.71-3.69 (m, 2H), 3.58-3.53 (m, 6H), 3.49-3.46 (m, 4H), 3.37-3.32 (m, 5H), 1.94-1.82 (m, 8H), 1.57-1.49 (m, 8H). ^{13}C NMR (126 MHz, CDCl_3), δ (ppm) 155.73, 153.21, 152.98, 150.01, 141.44, 123.81, 115.96, 115.21, 114.75, 114.04, 113.94, 72.05, 71.98, 71.05, 70.79, 70.65, 70.61, 70.16, 69.50, 68.61, 59.13, 34.02, 33.95, 32.83, 32.79, 29.38, 28.10, 26.83, 25.68, 25.48. HRMS (ESI) calcd for $\text{C}_{64}\text{H}_{87}\text{O}_{12}\text{N}_4\text{Br}^{81}\text{BrS}_4$, ($[\text{M}+\text{H}]^+$) 1551.1890, Found 1551.1884.

Compound **BGO6** (yield 42%). ^1H NMR (500 MHz, CDCl_3), δ (ppm) 7.25 (d, $J = 1.5$ Hz, 1H), 7.20 (d, $J = 1.7$ Hz, 1H), 6.92-6.91 (d, $J = 1.5$ Hz, 1H), 4.41-4.39 (t, $J = 6.5$ Hz, 2H), 4.10-4.04 (m, 4H), 3.70-3.68 (m, 2H), 3.59-3.56 (m, 2H), 3.53 (m, 2H), 3.51-3.34 (m, 11H), 1.94-1.84 (m, 8H), 1.57 (m, 8H). ^{13}C NMR (126 MHz, CDCl_3), δ (ppm) 156.49, 152.82, 149.59, 149.27, 146.38, 127.53, 118.70, 113.82, 113.78, 113.46, 111.50, 71.96, 71.16, 70.78, 70.64, 70.60, 70.12, 69.28, 69.06, 59.12, 45.14,

33.94, 32.84, 32.82, 29.25, 29.20, 28.06, 28.04, 25.41, 25.39. HRMS (ESI) calcd for $C_{64}H_{87}O_{12}N_4Br^{81}BrS_4$, ($[M+H]^+$) 1551.1890, Found 1551.1919.

Product **BGM6P**, **BGP6P** and **BGO6P** were prepared with the similar procedures as follow.

General procedures for **BGM6P**, **BGP6P** and **BGO6P**. **BGM6**, **BGP6** or **BGO6** (84 mg, 0.054 mmol) and sodium azide (50 mg, 0.75 mmol) were dissolved in DMF (10 mL), and the mixture was stirred for 3 h at room temperature. Then a large amount of water was added until all solids were dissolved. The reaction was extracted twice with ethyl acetate, the combined organic phase was dried with $MgSO_4$ and evaporated in vacuum. The crude product was subjected to flash column chromatography on silica gel to afford a dark green solid (69 mg, 0.052mmol). The dark green solid was dissolved in THF (5 mL) and copper (I) thiophene-2-carboxylate (CuTc) (5 mg), w-alkynyl-PEG-hydroxyl PEG1500 (M_n = 1500 mg/mL) (150 mg, about 0.1 mol) and tris[(1-benzyl-1H-1,2,3-triazol-4-yl)methyl]amine (TBTA) (3 mg) were added. The system was stirred at room temperature for 0.5 h, and then filtered with diatomite, and the solution was evaporated in vacuum. The crude product was purified by thin layer chromatography twice. First, ethyl acetate was used as an eluent and a small amount of impurities would move to the top of the TLC plate, but other parts of product remained at the start point of the TLC plate. Then DCM/MeOH (10:1-5:1) was used as an eluent successively, and the PEGylated product could be separated from alkyne-PEG (yiled ~85%).

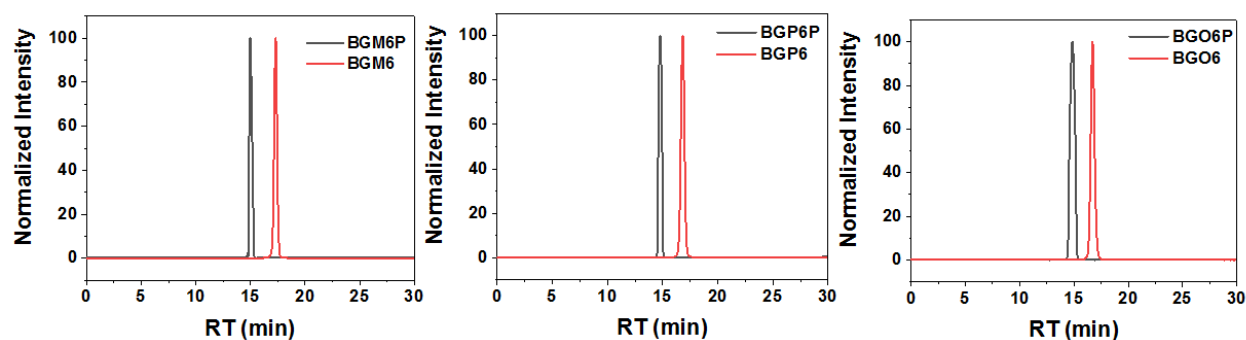
BGM6P (yield 86%). SEC measured: M_n = 7191 g/mol, M_w = 7676 g/mol, PDI = 1.067.

BGP6P (yield 83%). SEC measured: M_n = 7523 g/mol, M_w = 8006 g/mol, PDI = 1.064.

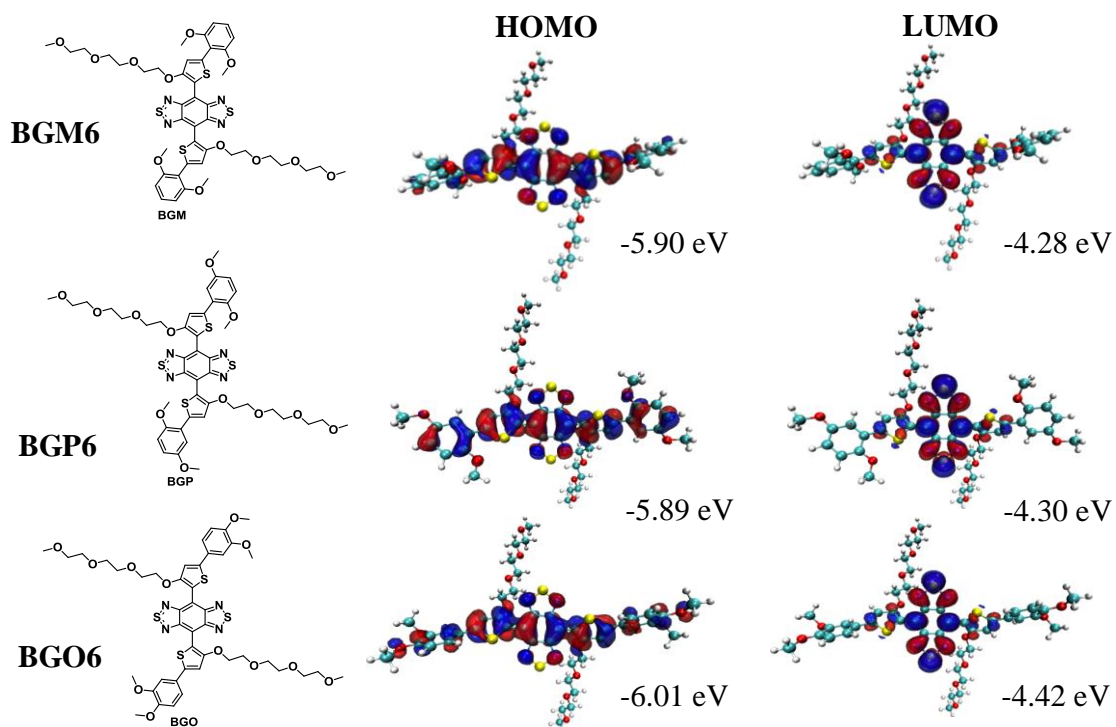
BGO6P (yield 89%). SEC measured: M_n = 7486 g/mol, M_w = 7953 g/mol, PDI = 1.063.

3 Supplementary Figures and Tables

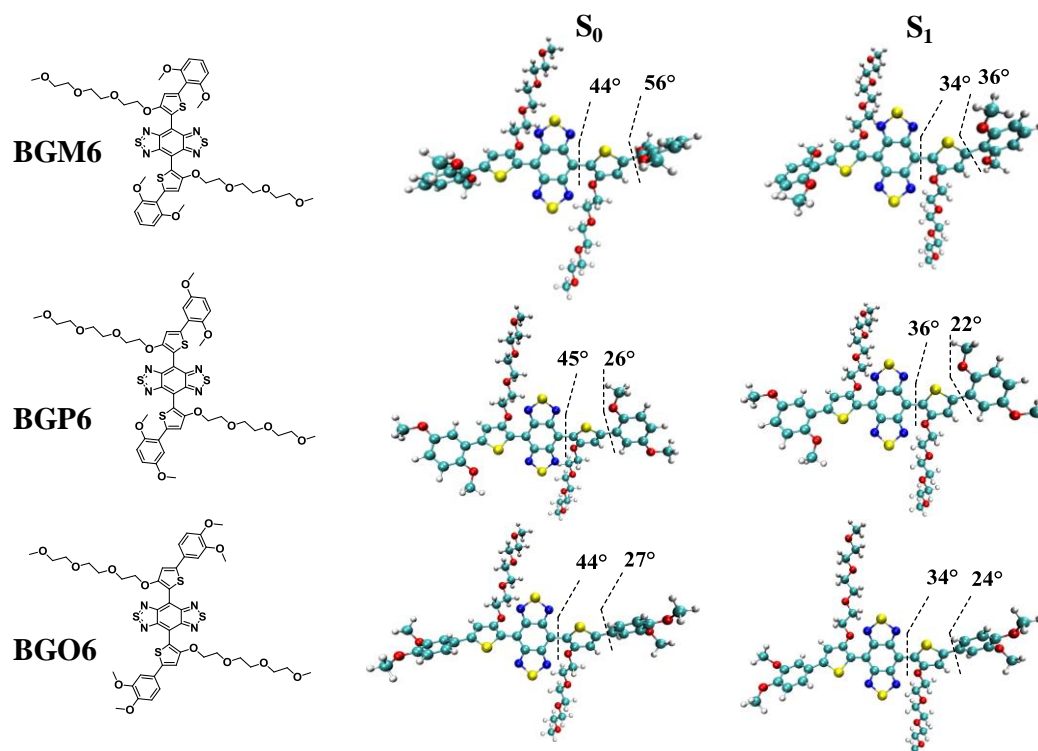
3.1 Supplementary Figures



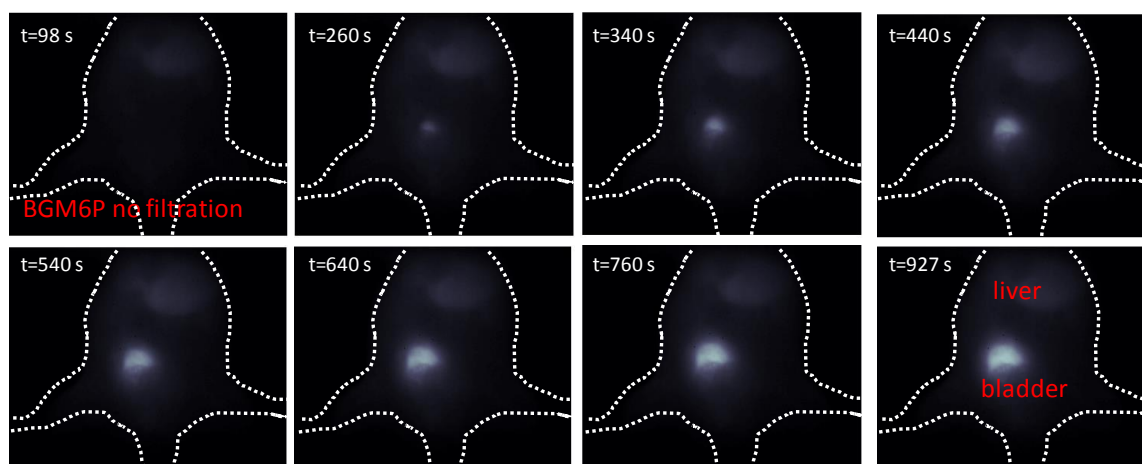
Supplementary Figure 1. Size-exclusion chromatography (SEC) analysis of the un-PEGylated and PEGylated molecular fluorophores.



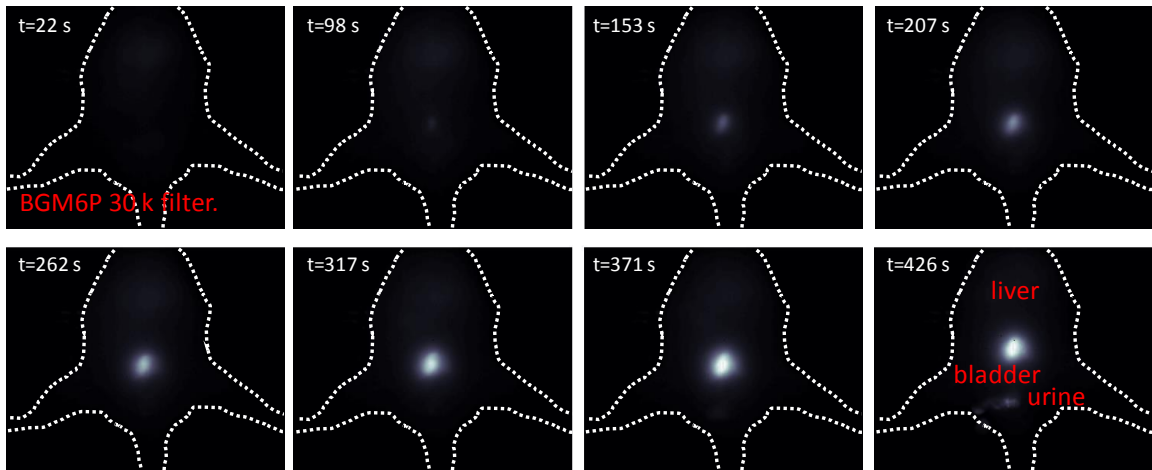
Supplementary Figure 2. Calculated HOMOs and LUMOs of the molecular fluorophores at the tuned- ω B97XD*/6-31G(d) level. The HOMO and LUMO energy levels are also presented in the figures. To reduce the computational requirements, side chains on the benzene units are replaced by methoxy groups. Note that the LUMO levels are obtained by subtracting the optical gap from the HOMO levels.



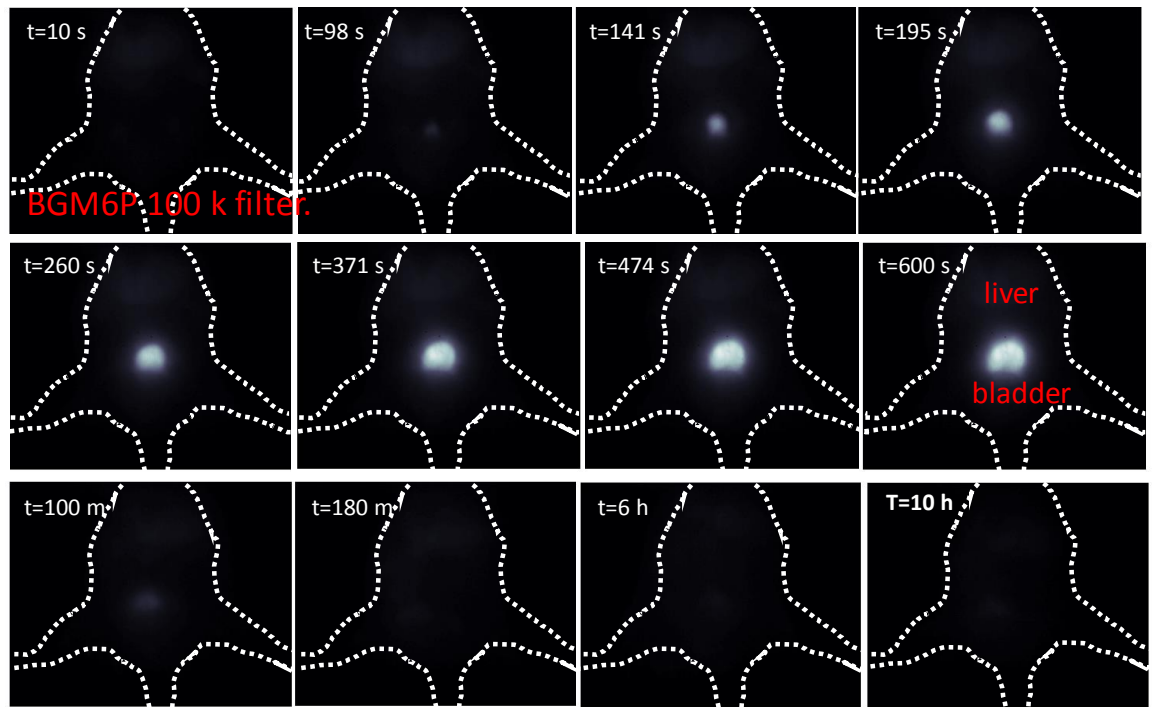
Supplementary Figure 3. Optimized ground-state (S_0) and first singlet excited state (S_1) geometries of the molecular fluorophores at the optimally tuned ω B97XD*/6-31G(d) level. To reduce the computational requirements, side chains on the fluorene units are replaced by methyloxy groups. The dihedral angles are shown.



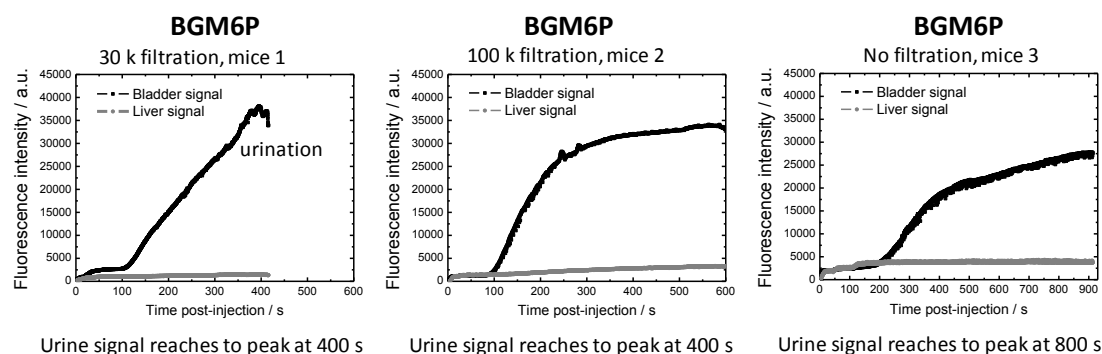
Supplementary Figure 4. Selected time points from video-rate NIR-II imaging of a mouse in the supine position after tail vein injection of as-prepared **BGM6P** (Inject dose: OD=2, 200 uL, 50 ms exposure time).



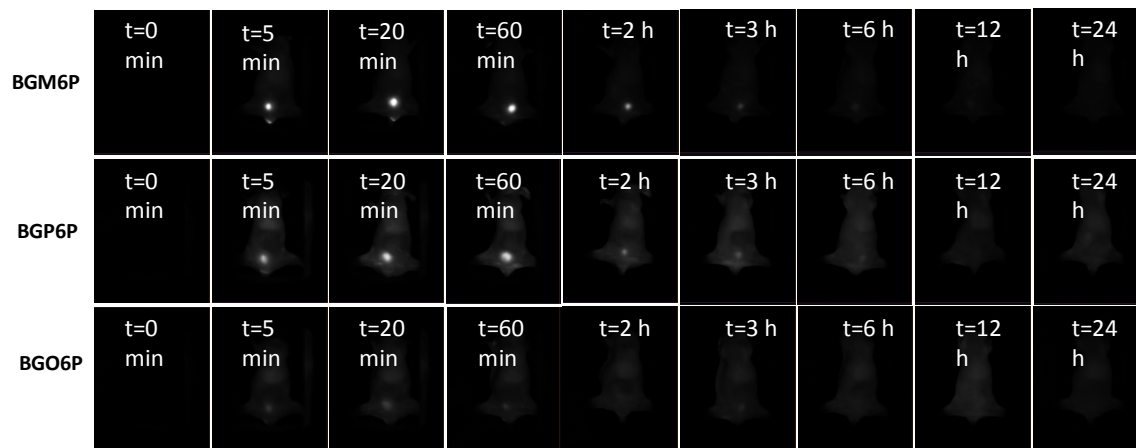
Supplementary Figure 5. Selected time points from video-rate NIR-II imaging of a mouse in the supine position after tail vein injection of 30 K filtered **BGM6P**.



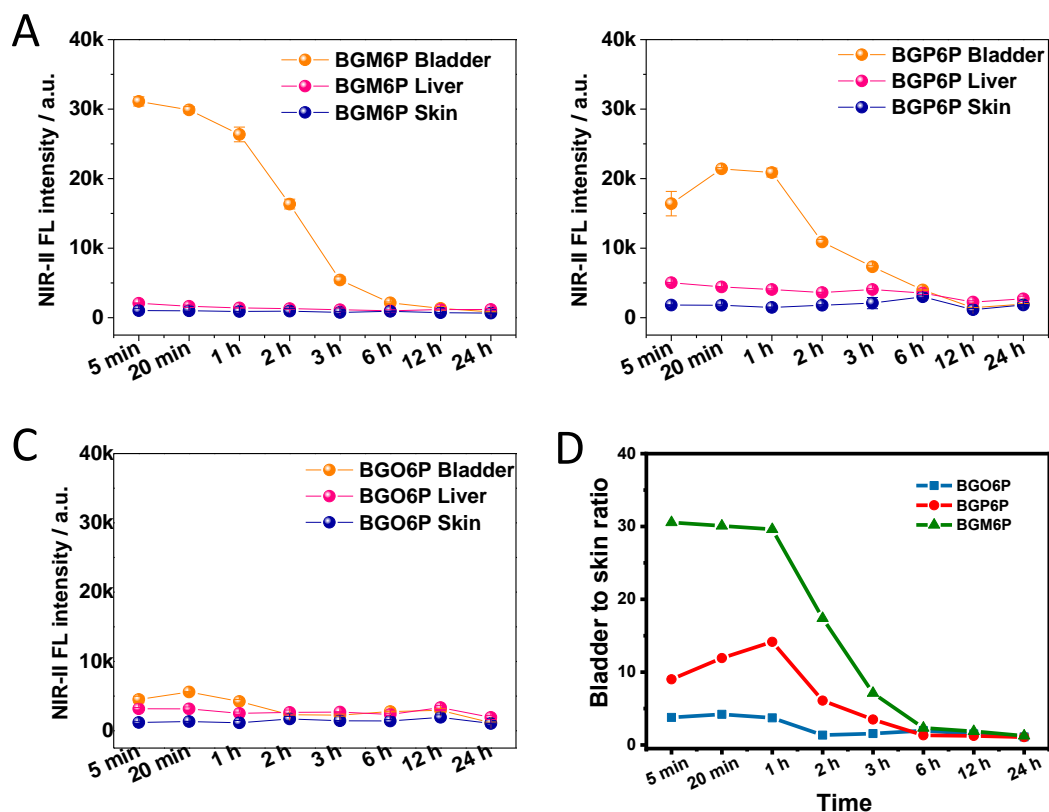
Supplementary Figure 6. Selected time points from video-rate NIR-II imaging of a mouse in the supine position after tail vein injection of 100 K filtered **BGM6P**.



Supplementary Figure 7. Bladder and liver signal of the mouse with an injection of 30 K filtered, 100 K filtered and as-prepared **BGM6P**. Liver signal is *ca.* 1000, 2500, 3500 counts, respectively.



Supplementary Figure 8. Selected time points from NIR-II whole body imaging of mice in the supine position after tail vein injection of 30 K filtrated **BGM6P**, **BGP6P** and **BGO6P**, respectively.



Supplementary Figure 9. NIR-II fluorescent signal intensity of liver (A), bladder (B) and skin (C) regions for **BGO6P**, **BGP6P** and **BGM6P** injected mouse at different time after injection. (D) Representative background subtracted signal of bladder to skin as a function of time for mice injected with **BGM6P**, **BGP6P** and **BGO6P**, respectively.

Chemical structure of 2,2'-(1,3-phenylene)-5,5'-bis(4-bromophenyl) ether:

BrC1=CC=C(C=C1)Oc2cc3ccccc3cc2Oc4cc(Br)ccc4

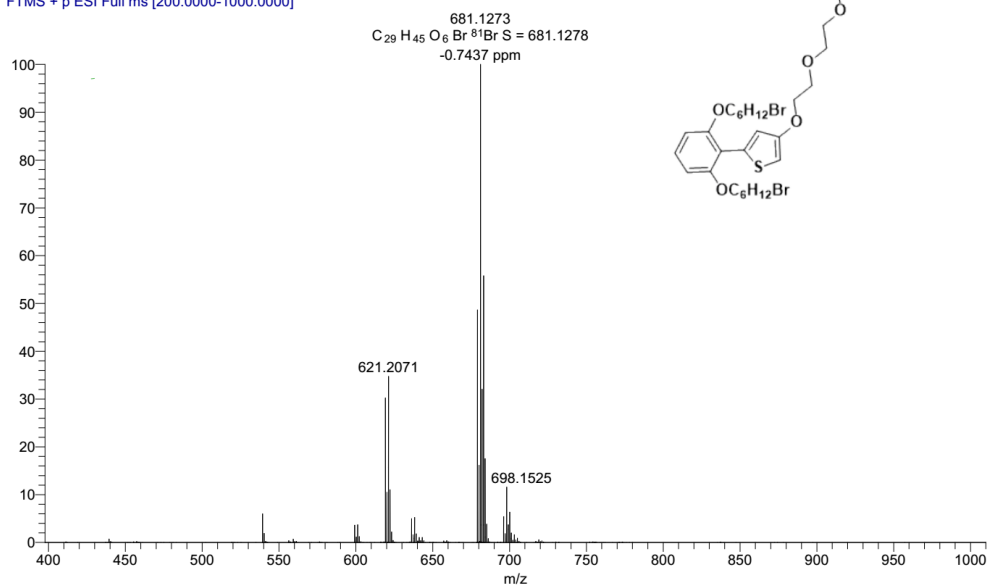
¹³C NMR spectrum (CDCl₃) showing peaks at the following chemical shifts (ppm):

- 157.35
- 156.11
- 133.25
- 128.65
- 121.22
- 113.07
- 105.46
- 97.85
- 72.08
- 70.91
- 70.81
- 70.72
- 69.94
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- 68.98
- 59.18
- 33.99
- 32.80
- 29.07
- 27.94
- 25.46

12

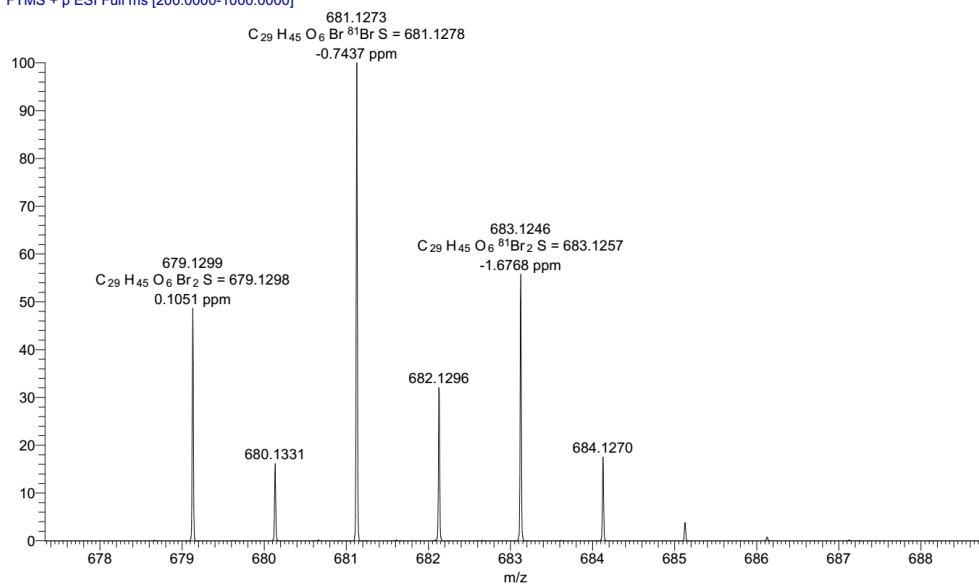
Positive:

14 #17 RT: 0.16 AV: 1 NL: 4.35E9
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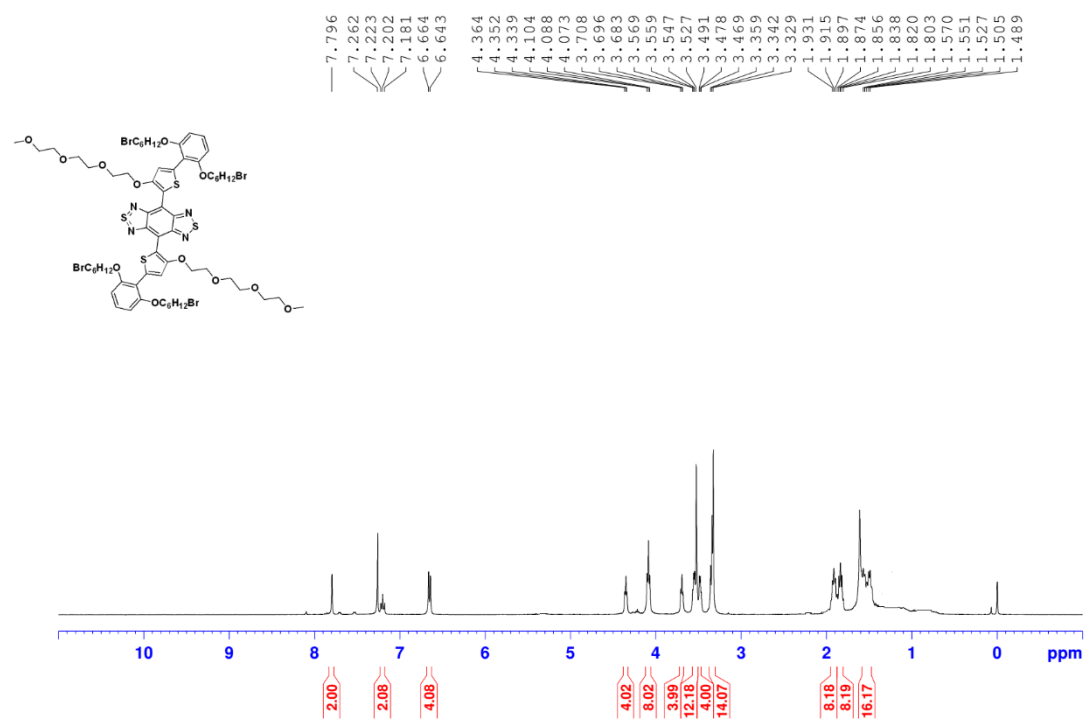
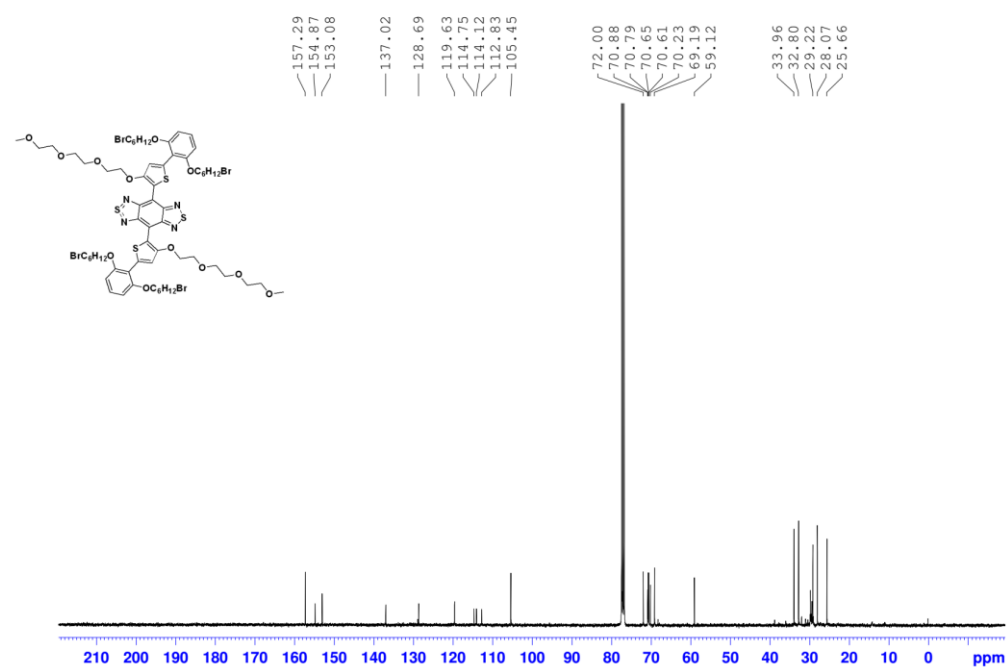


[M+H]

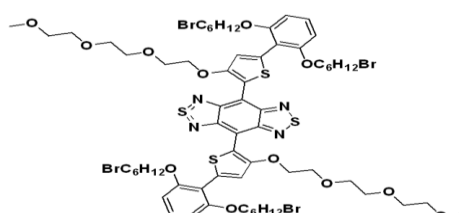
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T: FTMS + p ESI Full ms [200.0000-1000.0000]



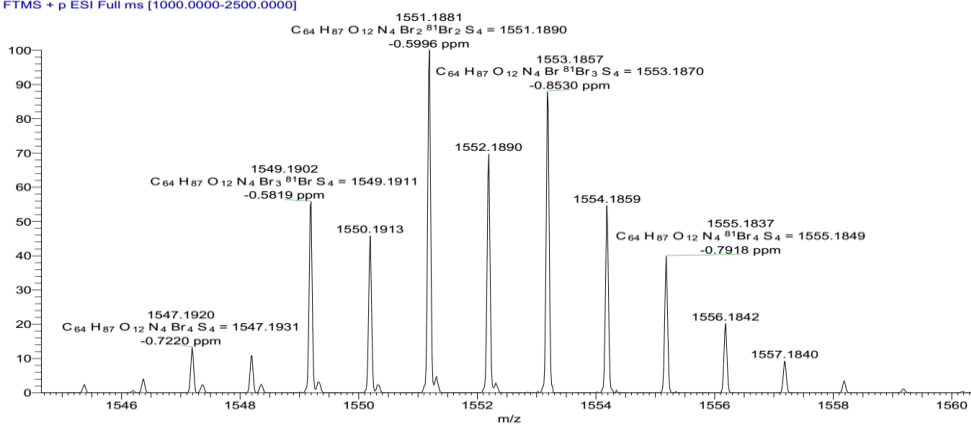
Supplementary Figure 12. HR MASS spectrum of M1.

Supplementary Figure 13. ^1H NMR spectrum of BGM6.Supplementary Figure 14. ^{13}C NMR spectrum of BGM6.

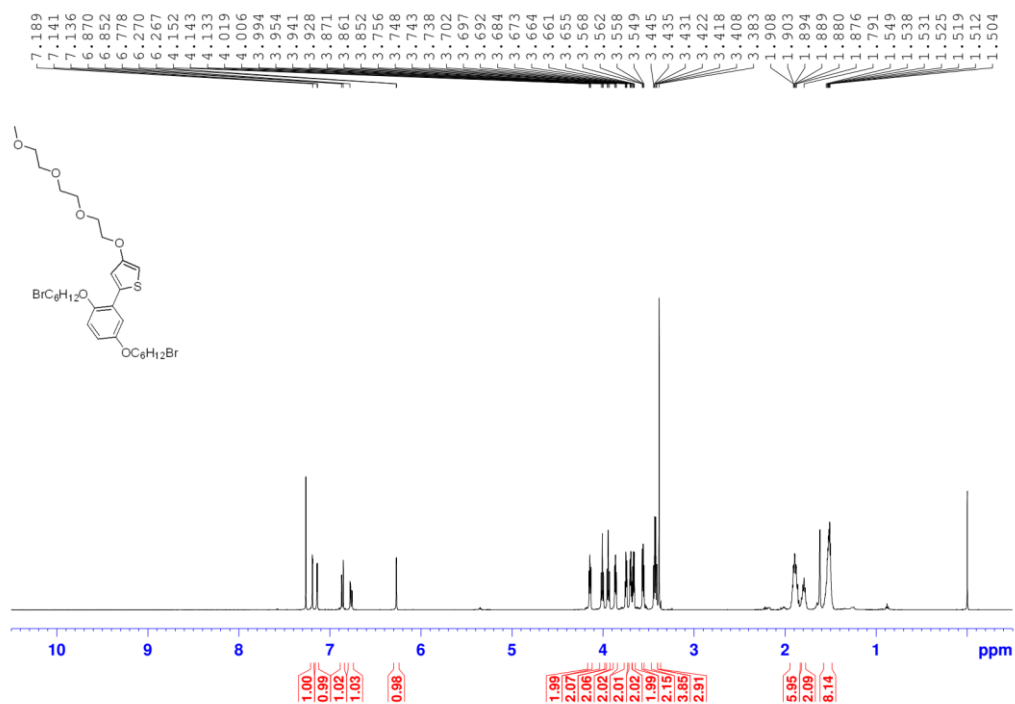
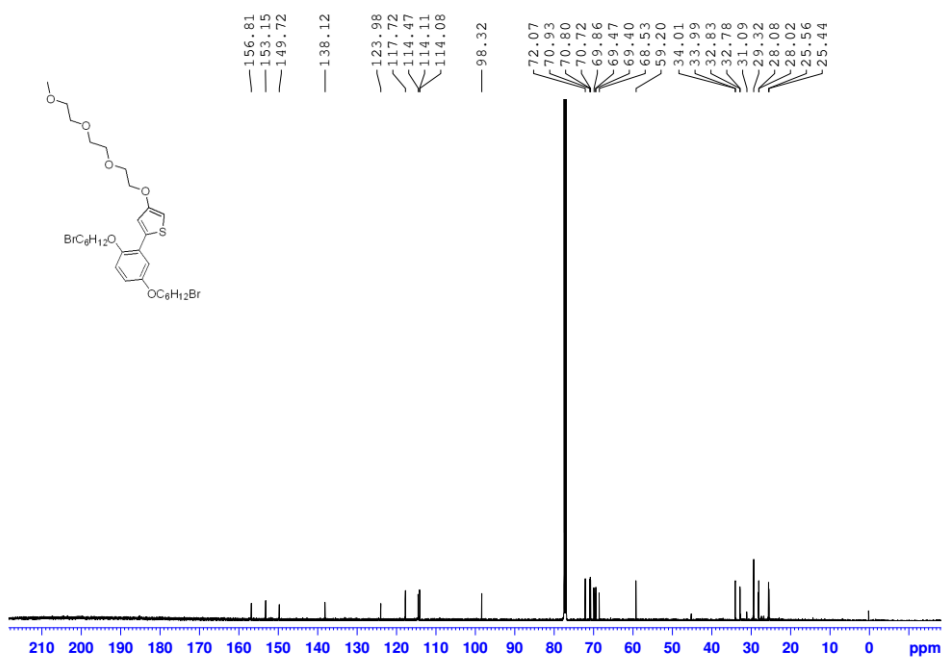
LM3-Br-d2_170425174628 #7 RT: 0.08 AV: 1 NL: 2.26E7
T: FTMS +p ESI Full ms [1000.0000-2500.0000]



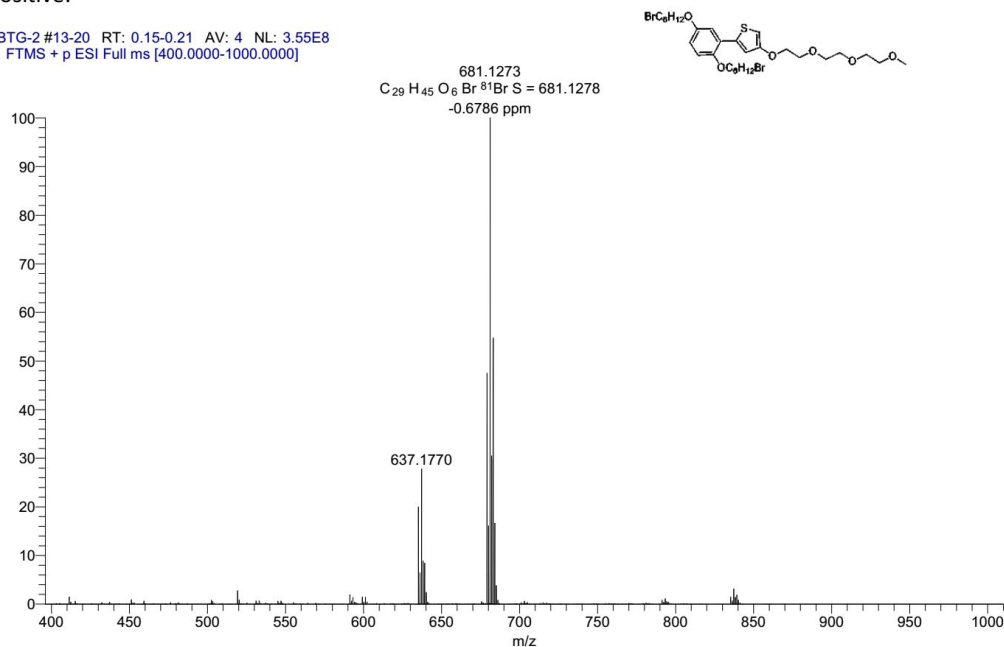
LM3-Br-d2_170425174628 #7 RT: 0.08 AV: 1 NL: 2.26E7
T: FTMS + p ESI Full ms [1000.0000-2500.0000]



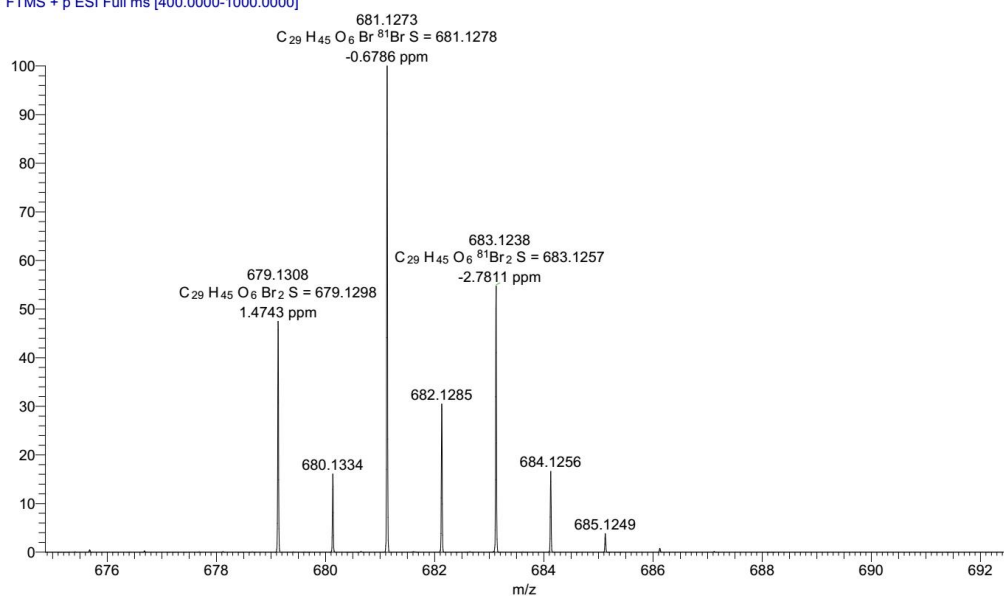
Supplementary Figure 15. HR MASS spectrum of **BGM6**.

Supplementary Figure 16. ¹H NMR spectrum of M2.Supplementary Figure 17. ¹³C NMR spectrum of M2.

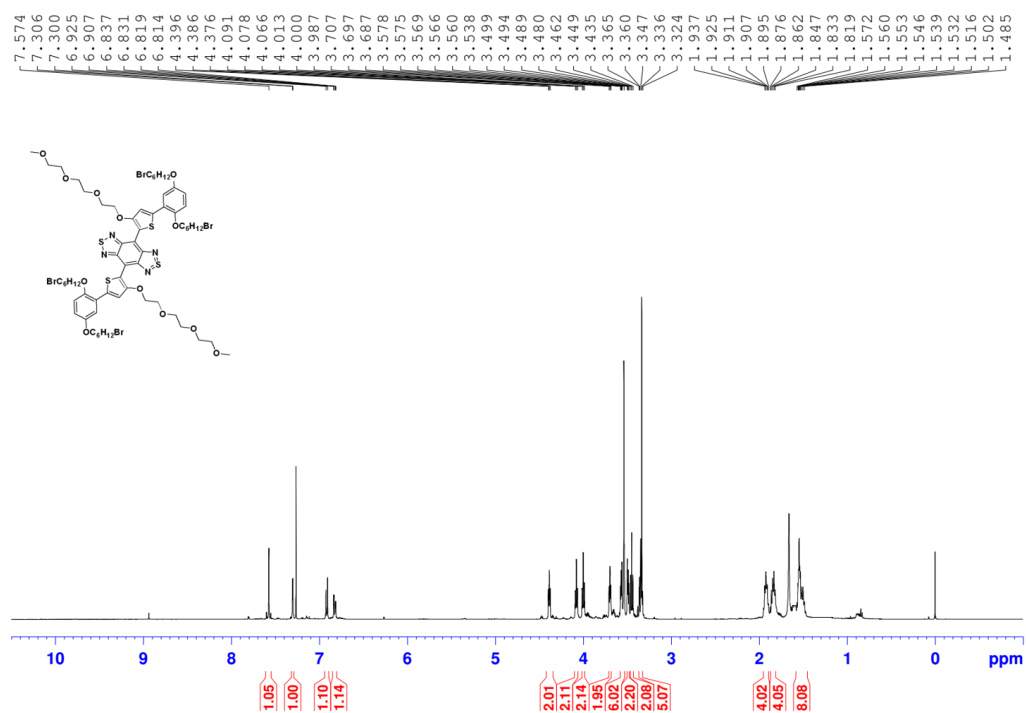
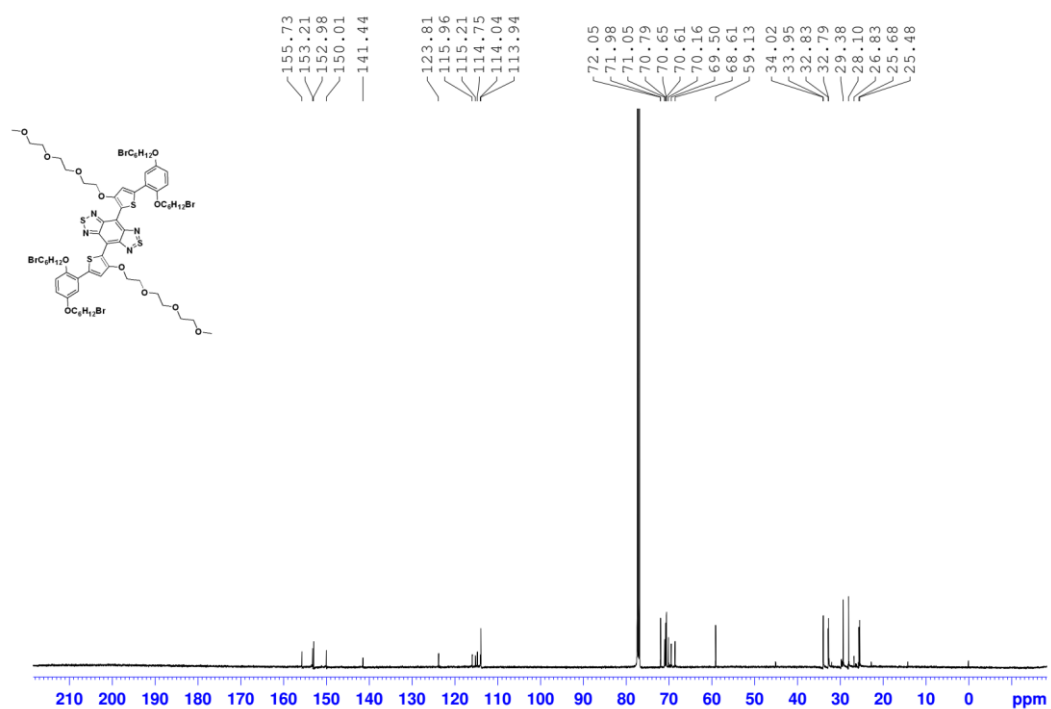
PBTG-2 #13-20 RT: 0.15-0.21 AV: 4 NL: 3.55E8
T: FTMS + p ESI Full ms [400.0000-1000.0000]



PBTG-2 #13-20 RT: 0.15-0.21 AV: 4 NL: 3.55E8
T: FTMS + p ESI Full ms [400.0000-1000.0000]

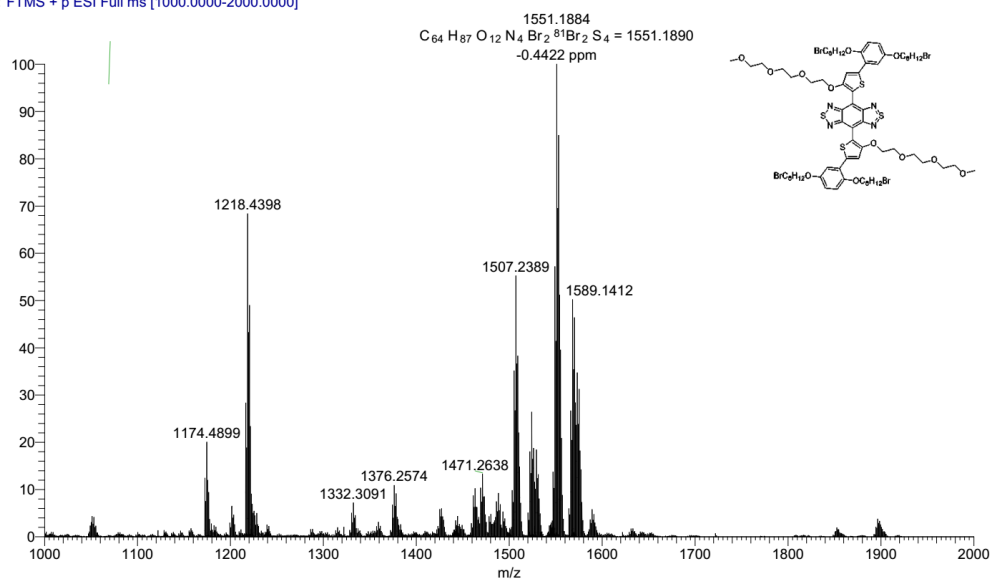


Supplementary Figure 18. HR MASS spectrum of M2.

Supplementary Figure 19. ^1H NMR spectrum of BGP6.Supplementary Figure 20. ^{13}C NMR spectrum of BGP6.

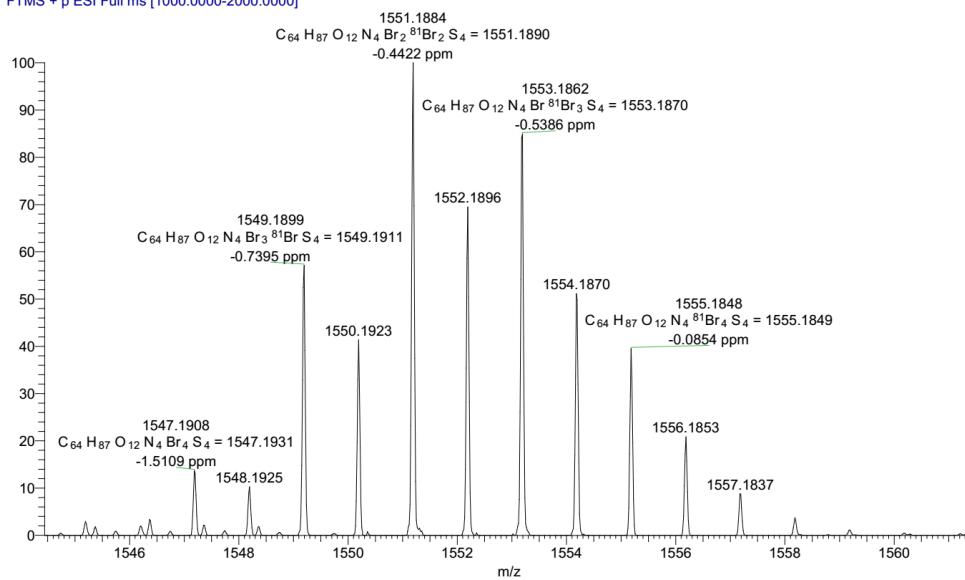
Positive:

LM3-P-Br-d2_170731154508 #34 RT: 0.43 AV: 1 NL: 1.91E6
T: FTMS + p ESI Full ms [1000.0000-2000.0000]

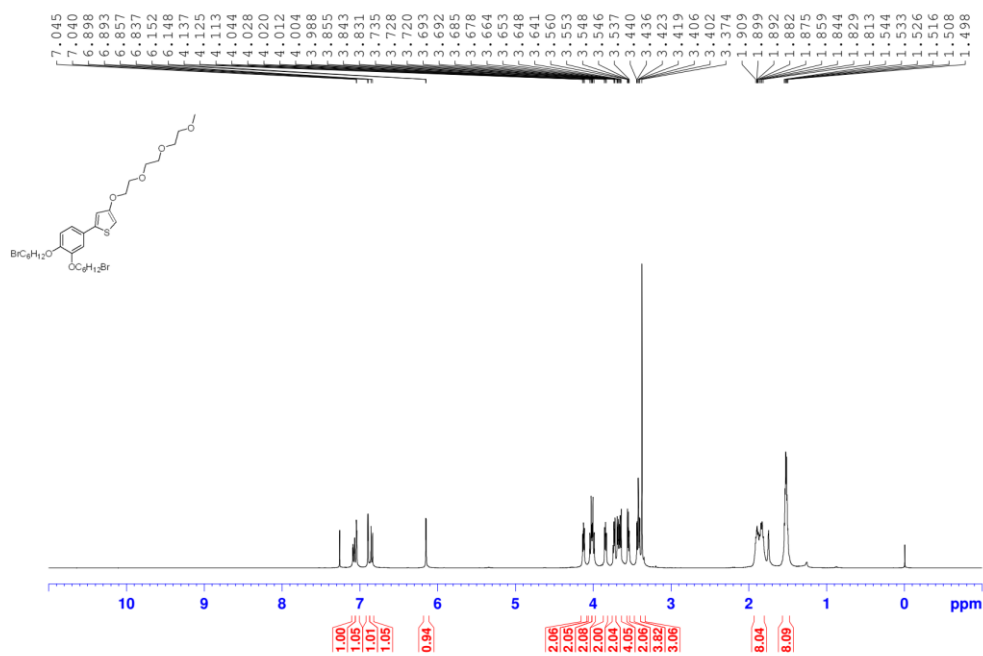


[M+H]

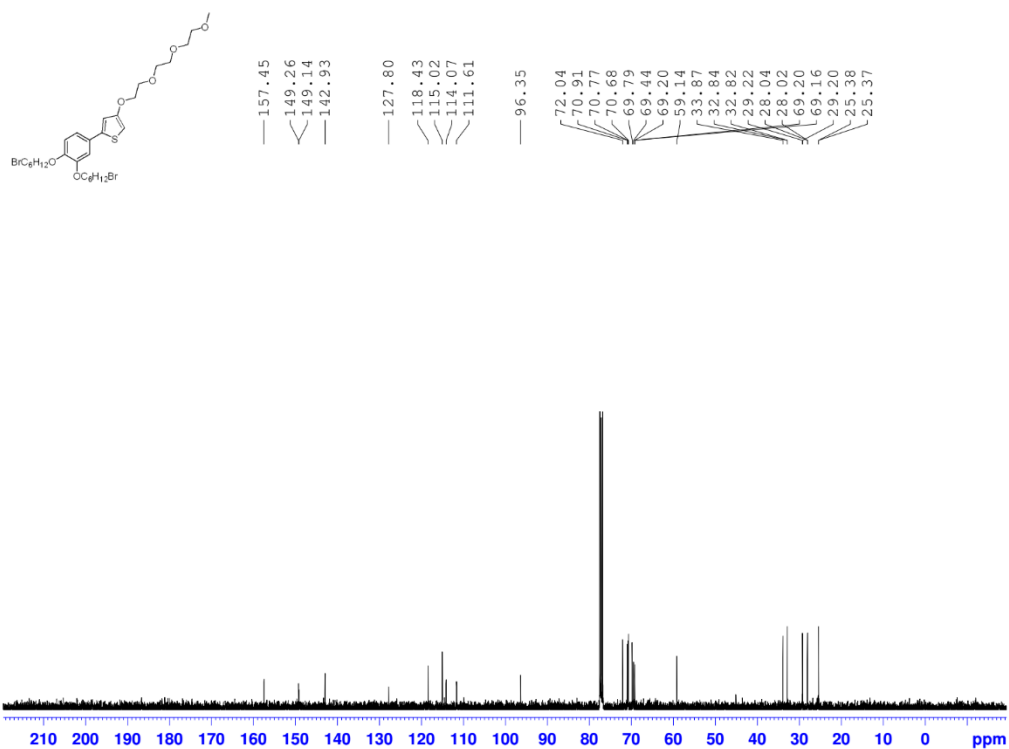
LM3-P-Br-d2_170731154508 #34 RT: 0.43 AV: 1 NL: 1.91E6
T: FTMS + p ESI Full ms [1000.0000-2000.0000]



Supplementary Figure 21. HR MASS spectrum of BGP6.

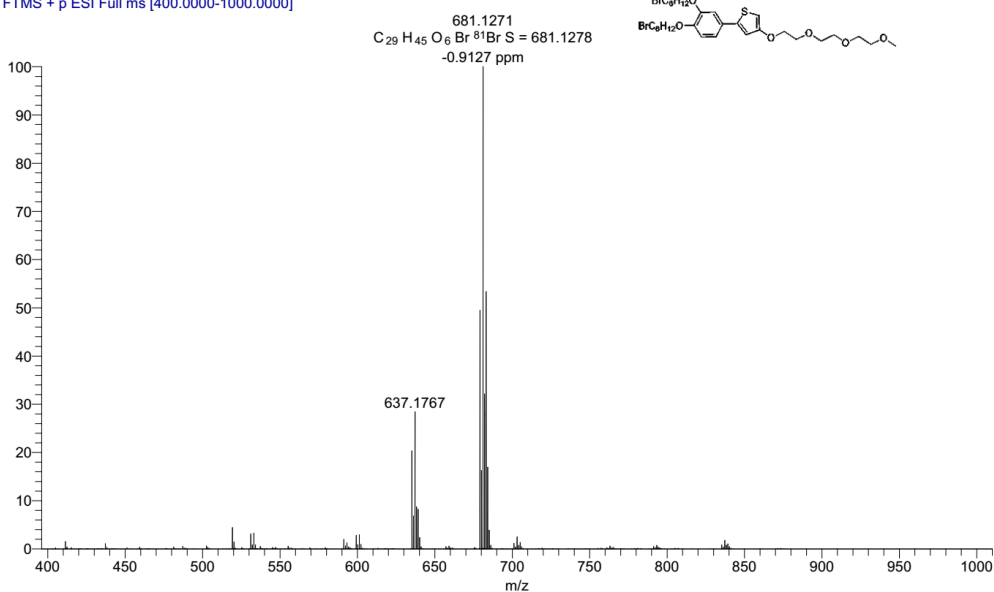


Supplementary Figure 22. ^1H NMR spectrum of M3.

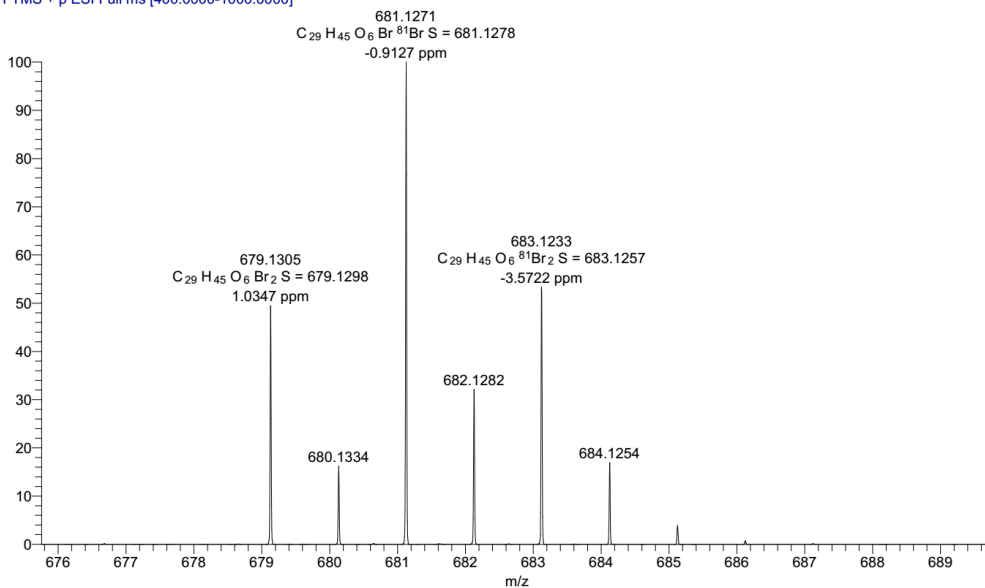


Supplementary Figure 23. ^{13}C NMR spectrum of M3.

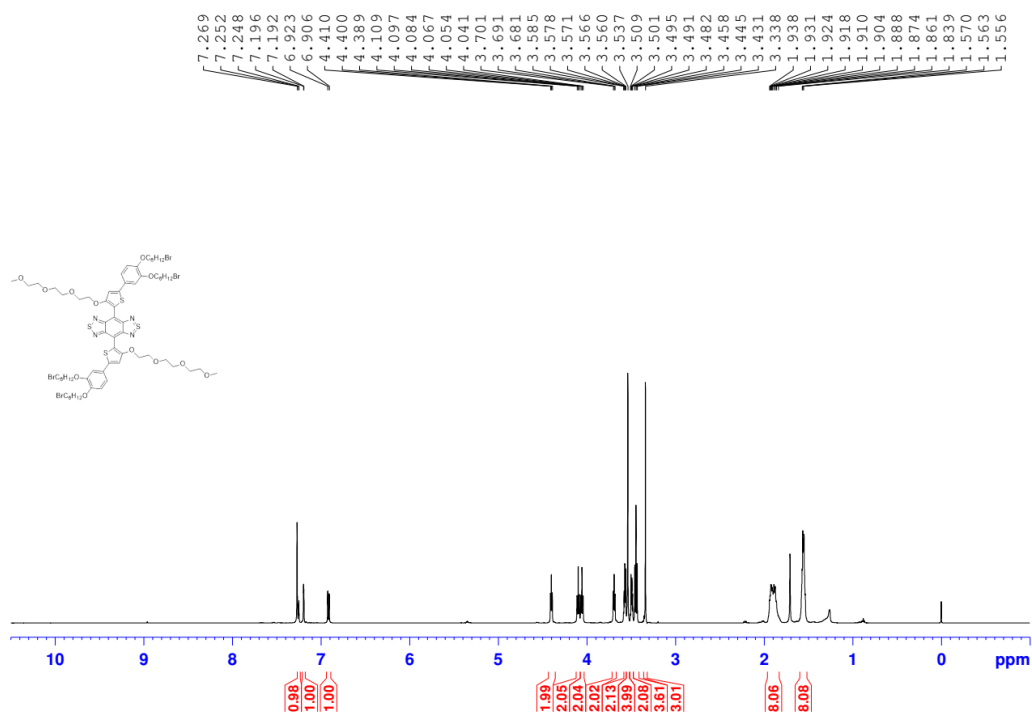
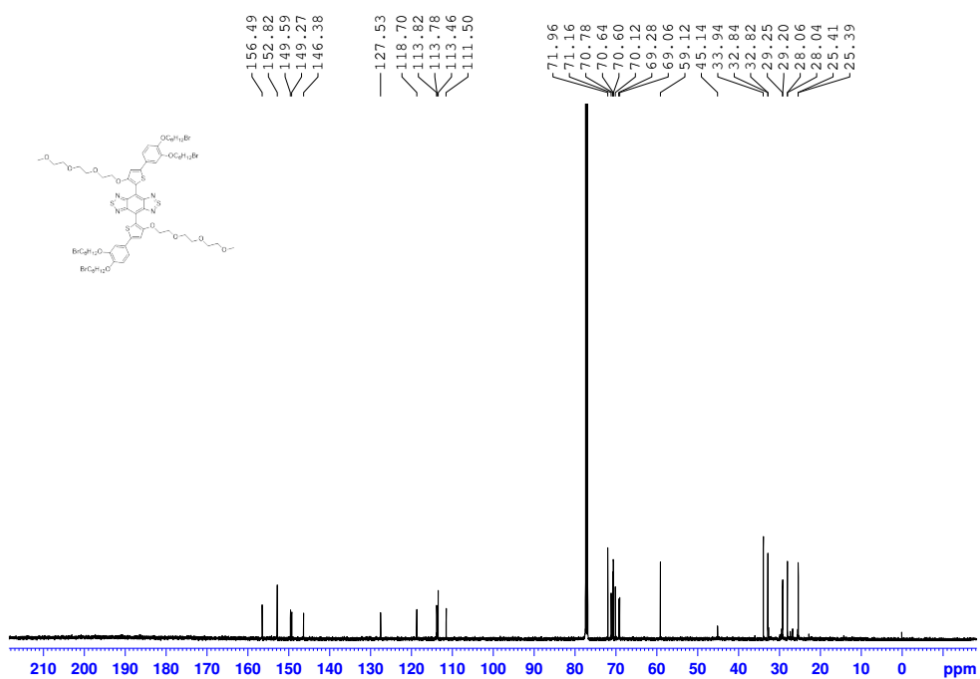
OBTG-2 #13-23 RT: 0.15-0.22 AV: 5 NL: 3.69E8
T: FTMS + p ESI Full ms [400.0000-1000.0000]



OBTG-2 #13-23 RT: 0.15-0.22 AV: 5 NL: 3.69E8
T: FTMS + p ESI Full ms [400.0000-1000.0000]

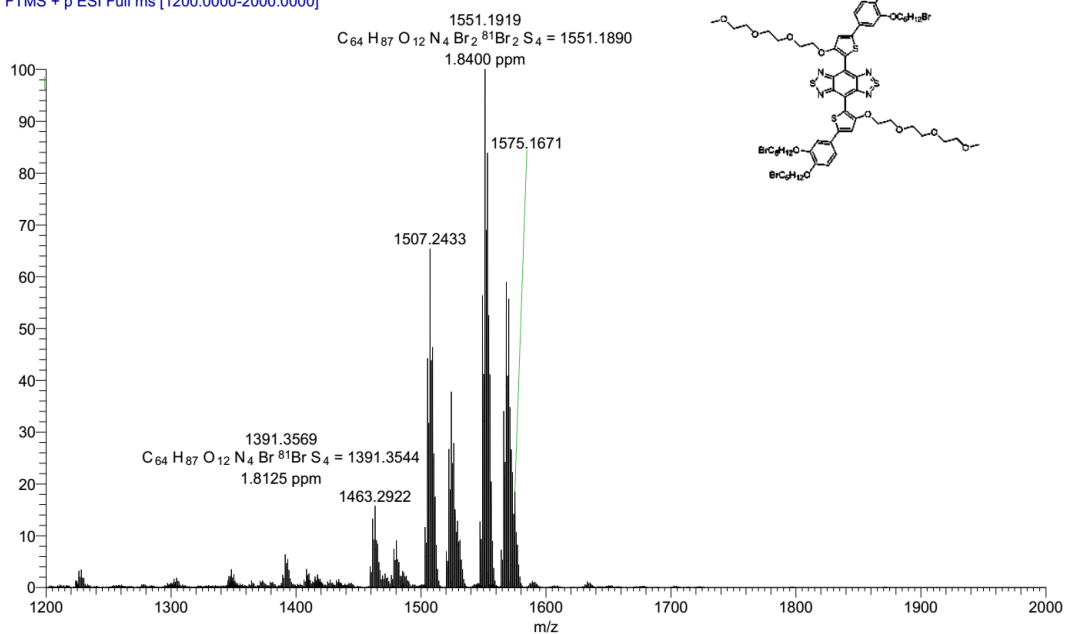


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Supplementary Figure 25. ¹H NMR spectrum of BGO6.Supplementary Figure 26. ¹³C NMR spectrum of BGO6.

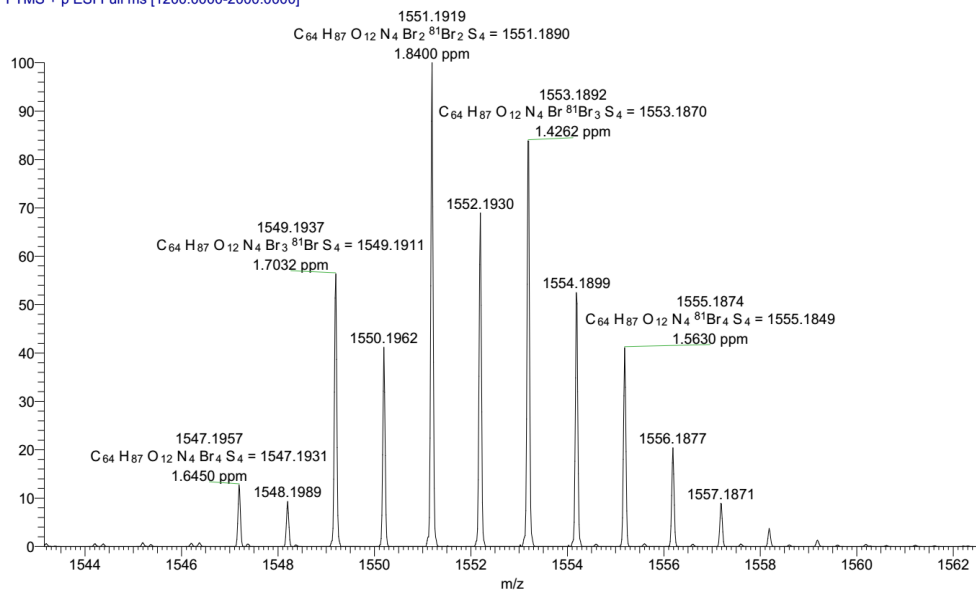
Positive:

LM3-O-Br_170822172050 #23 RT: 0.29 AV: 1 NL: 8.43E6
T: FTMS + p ESI Full ms [1200.0000-2000.0000]



[M+H]

LM3-O-Br_170822172050 #23 RT: 0.29 AV: 1 NL: 8.43E6
T: FTMS + p ESI Full ms [1200.0000-2000.0000]



Supplementary Figure 27. HR MASS spectrum of BGO6.

Supplementary Table 1. Calculated first vertical S_0 - S_1 excitation energies (E_{01}), first vertical S_1 - S_0 emission energies (E_{10}), electronic configurations determined at the TD- ω B97XD*/6-31G (d) level of theory. ^aThe optimally tuned range-separated parameters included in the functionals. ^{ex}Experimental data.

Molecules	ω^* ^a	E_{01} (λ_{01}) ev (nm)	f_{01}	Electronic configuration	E_{10} (λ_{10}) ev (nm)	f_{10}	λ_{01}^{ex} (nm)	λ_{10}^{ex} (nm)
BGM6	0.1182	1.62(764)	0.32	HOMO \rightarrow LUMO 99%	1.14(1092)	0.32	736	1047
BGP6	0.1182	1.59(780)	0.39	HOMO \rightarrow LUMO 98%	1.13(1094)	0.36	736	1060
BGO6	0.1187	1.59(781)	0.39	HOMO \rightarrow LUMO 98%	1.13(1093)	0.36	741	1060

Supplementary Table 2. Size-exclusion chromatography (SEC) analysis of the NIR-II molecular fluorophores.

Fluorophores	RV (mL)	M_n (Daltons)	M_w (Daltons)	M_w/M_n
BGM6P	14.97	7191	7676	1.067
BGP6P	14.78	7523	8006	1.064
BGO6P	14.85	7486	7953	1.063

Supplementary Table 3. Pass percentage analysis of the fluorophores with different filters.

	Filtration	OD at peak	Pass percentage (%)
BGM6P	No filtration	3.91	100.00
	100k filtration	3.64	93.10
	50k filtration	0.98	25.10
	30k filtration	0.04	1.08
BGP6P	No filtration	3.83	100.00
	100k filtration	3.23	84.33
	50k filtration	0.36	9.40
	30k filtration	0.00	0.00
BGO6P	No filtration	3.77	100.00
	100k filtration	2.32	61.53
	50k filtration	0.14	3.66
	30k filtration	0.00	0.00

4 Reference

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