

Supplementary Material

1 Supplementary Data

Mass isotopomer-balance equations for C8:0 :

$$\frac{d}{dt} [C8:0_0] = \frac{1}{C8:0} \left(F_5 [C8:0_0^{medium}] - F_5^{inv} [C8:0_8] - F_6 [C8:0_0] + F_6^{inv} [3OHC8:0_0] \right)$$

$$\frac{d}{dt} [3OHC8:0_0] = \frac{1}{3OHC8:0} \left(F_6 [C8:0_0] - F_6^{inv} [3OHC8:0_8] - F_8 [3OHC8:0_0] + F_8^{inv} [C6:0_0 \cdot AcCoA_0] - F_7 [3OHC8:0_0] \right)$$

$$\frac{d}{dt} [C6:0_0] = \frac{1}{C6:0} \left(F_8 [3OHC8:0_0] - F_8^{inv} [C6:0_6] - F_{10} [C6:0_0] + F_{10}^{inv} [3OHC6:0_0] - F_9 [C6:0_0] \right)$$

$$\frac{d}{dt} [3OHC6:0_0] = \frac{1}{3OHC6:0} \left(F_{10} [C6:0_0] - F_{10}^{inv} [3OHC6:0_6] - F_{12} [3OHC6:0_0] + F_{12}^{inv} [AcCoA_0 \cdot AcCoA_0 \cdot AcCoA_0] - F_{11} [3OHC6:0_0] \right)$$

$$\frac{d}{dt} [AcCoA_0] = \frac{1}{AcCoA} \left(F_8 [3OHC8:0_0] - F_8^{inv} [AcCoA_2] + F_{12} [3OHC6:0_0] - F_{12}^{inv} [AcCoA_2 \cdot AcCoA_2 \cdot AcCoA_2] - F_{14} [AcCoA_0 \cdot AcCoA_0] + F_{13} [1] \right)$$

Mass isotopomer-balance equations for C10:0:

$$\frac{d}{dt} [C10:0_0] = \frac{1}{C10:0} \left(F_1 [C10:0_0^{medium}] - F_1^{inv} [C10:0_{10}] - F_2 [C10:0_0] + F_2^{inv} [3OHC10:0_0] \right)$$

$$\frac{d}{dt} [3OHC10:0_0] = \frac{1}{3OHC10:0} \left(F_2 [C10:0_0] - F_2^{inv} [3OHC10:0_{10}] - F_4 [3OHC10:0_0] + F_4^{inv} [C8:0_0 \cdot AcCoA_0] - F_3 [3OHC10:0_0] \right)$$

$$\frac{d}{dt} [C8:0_0] = \frac{1}{C8:0} \left(F_4 [3OHC10:0_0] - F_4^{inv} [C8:0_8] - F_6 [C8:0_0] + F_6^{inv} [3OHC8:0_0] - F_5^{inv} [C8:0_0] \right)$$

$$\frac{d}{dt} [3OHC8:0_0] = \frac{1}{3OHC8:0} \left(F_6 [C8:0_0] - F_6^{inv} [3OHC8:0_8] - F_8 [3OHC8:0_0] + F_8^{inv} [C6:0_0 \cdot AcCoA_0] - F_7 [3OHC8:0_0] \right)$$

$$\frac{d}{dt} [C6:0_0] = \frac{1}{C6:0} \left(F_8 [3OHC8:0_0] - F_8^{inv} [C6:0_6] - F_{10} [C6:0_0] + F_{10}^{inv} [3OHC6:0_0] - F_9 [C6:0_0] \right)$$

$$\frac{d}{dt} [3OHC6:0_0] = \frac{1}{3OHC6:0} \left(F_{10} [C6:0_0] - F_{10}^{inv} [3OHC6:0_6] - F_{12} [3OHC6:0_0] + F_{12}^{inv} [AcCoA_0 \cdot AcCoA_0 \cdot AcCoA_0] - F_{11} [3OHC6:0_0] \right)$$

$$\frac{d}{dt} \begin{bmatrix} \text{AcCoA}_0 \\ \text{AcCoA}_2 \end{bmatrix} = \frac{1}{\text{AcCoA}} \left(F_4 \begin{bmatrix} 3\text{OHC10:0}_0 \\ 3\text{OHC10:0}_{10} \end{bmatrix} - F_4^{\text{inv}} \begin{bmatrix} \text{AcCoA}_0 \\ \text{AcCoA}_2 \end{bmatrix} + F_8 \begin{bmatrix} 3\text{OHC8:0}_0 \\ 3\text{OHC8:0}_8 \end{bmatrix} - F_8^{\text{inv}} \begin{bmatrix} \text{AcCoA}_0 \\ \text{AcCoA}_2 \end{bmatrix} + F_{12} \begin{bmatrix} 3\text{OHC6:0}_0 \\ 3\text{OHC6:0}_6 \end{bmatrix} \right. \\ \left. - F_{12}^{\text{inv}} \begin{bmatrix} \text{AcCoA}_0 \cdot \text{AcCoA}_0 \cdot \text{AcCoA}_0 \\ \text{AcCoA}_2 \cdot \text{AcCoA}_2 \cdot \text{AcCoA}_2 \end{bmatrix} - F_{14} \begin{bmatrix} \text{AcCoA}_0 \cdot \text{AcCoA}_0 \\ \text{AcCoA}_2 \cdot \text{AcCoA}_2 \end{bmatrix} + F_{13} \begin{bmatrix} 1 \\ 0 \end{bmatrix} \right)$$

2 Supplementary Table

	C8:0		C10:0	
	flux	SD	flux	SD
F ₁	NA	NA	0.217	0.069
F ₁ ^{inv}	NA	NA	0.236	0.077
F ₂	NA	NA	0.012	0.004
F ₂ ^{inv}	NA	NA	0.027	0.010
F ₄	NA	NA	0.020	0.014
F ₄ ^{inv}	NA	NA	0.045	0.015
F ₅	0.135	0.029	NA	NA
F ₅ ^{inv}	0.080	0.031	NA	NA
F ₆	0.098	0.014	0.091	0.131
F ₆ ^{inv}	0.043	0.011	0.238	0.129**
F ₈	0.140	0.030	0.000	0.000***
F ₈ ^{inv}	0.109	0.027	0.221	0.074**
F ₁₀	0.532	0.379	0.000	0.000**
F ₁₀ ^{inv}	0.877	0.452	0.241	0.057**
F ₁₂	0.000	0.000	0.708	0.666*
F ₁₂ ^{inv}	0.147	0.031	0.930	0.644**
F ₁₃	0.187	0.017	0.490	0.045***
Calculated differences				
F ₆ - F ₆ ^{inv}	0.056	0.018	-0.147	0.184**
F ₈ - F ₈ ^{inv}	0.031	0.041	-0.221	0.074*
F ₁₀ - F ₁₀ ^{inv}	-0.345	0.590	-0.241	0.057
F ₁₂ - F ₁₂ ^{inv}	-0.147	0.031	-0.222	0.926

Table S1 : Fluxes (in $\mu\text{mol/g prot/min}$) resulting from the best fit of the mathematical model of β -oxidation. NA : non-available. * $p < 0.05$; ** $p < 0.01$; *** $p < 0.001$

3 Supplementary Figure

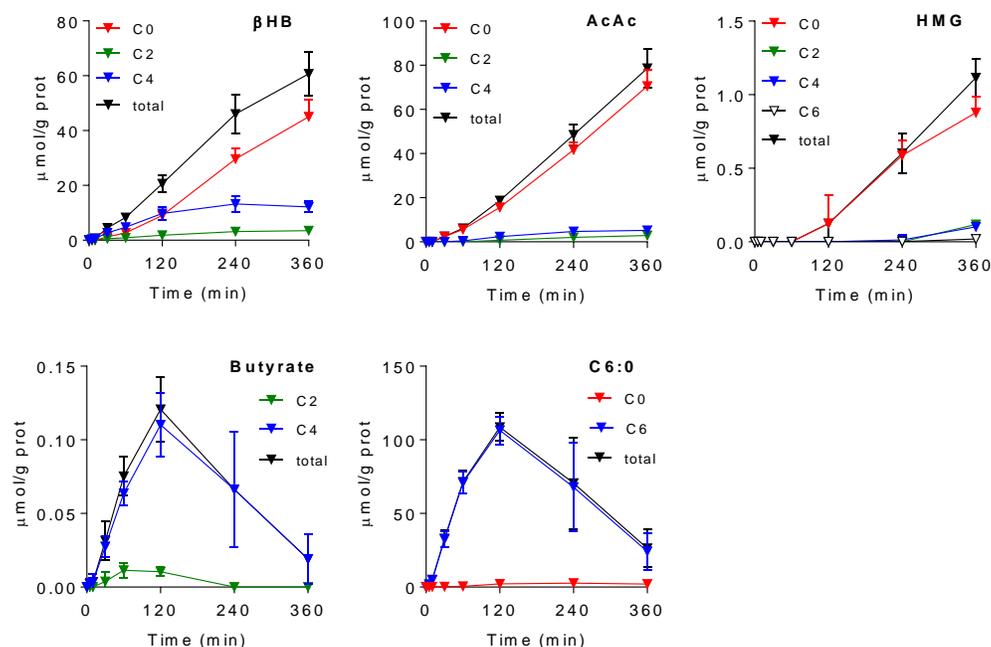


Figure S1 : Secretion rates of selected ketone bodies, HMG, butyrate and C6:0 determined from the total extracellular concentrations experimentally measured in HepG2 cells incubated with $[\text{U-}^{13}\text{C}]\text{-C8:0}$ (mean \pm SD, n=3). Secretion of βHB and AcAc was detected at time 5 min and 30 min, respectively, consistent with the general observation that the liver is the major organ for ketone production (Evans et al., 2017). Secretion of HMG, butyrate and C6:0 was detected at 120 min, 5 min and 5 min, respectively. Detection of mass isotopomers for HMG and butyrate was different as compared to iPSC astrocytes. Concentrations of butyrate and C6:0 peaked at 120 min and then decreased (inverted U-shape). All together, these data show different metabolite quantification in different cell types under similar conditions, supporting the sensitivity and the specificity of the measurements.