

Non-Halogenated Ionic Liquid Dramatically Enhances Tribological Performance of Biodegradable Oils

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1. Viscosity data

Table S1: Properties of the tested lubricants at atmospheric pressure

Fluid	Fluid Dynamic Viscosity [mPa s]	
	40°C	100°C
ME	3.1 ± 0.1	1.3 ± 0.2
ME / IL	3.5 ± 0.1	1.7 ± 0.2
OPAG	43.0 ± 0.2	7.7 ± 0.3
OPAG / IL	52.6 ± 0.2	9.2 ± 0.1

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2. Ball-on-three plates experimental setup

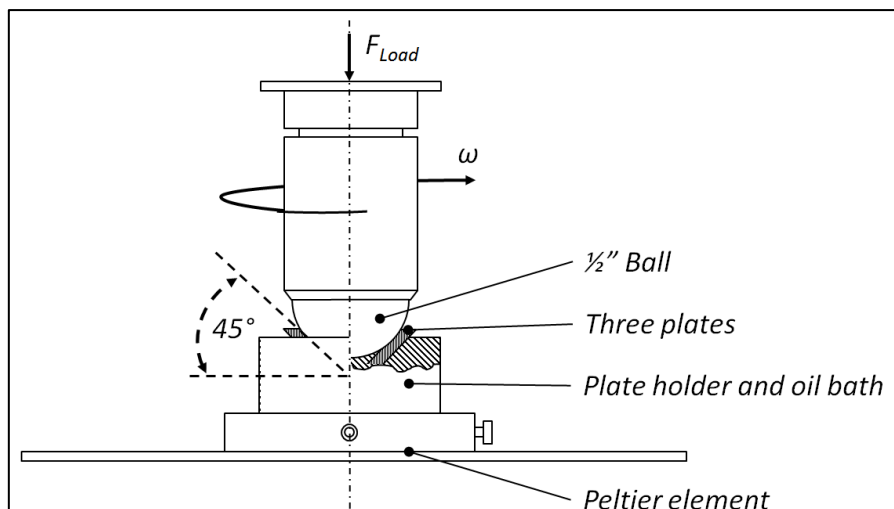


Figure S1: Ball-on-three-plates experimental setup

3. Friction data

Normal and friction forces were recorded with an acquisition frequency of 1 Hz. In the beginning of each test a running-in process resulted in a large variation of the coefficient of friction. The variation decreased with time in all cases but returned for the case ME + IL. The addition of IL to OPAG clearly reduced the friction compared to the neat oil.

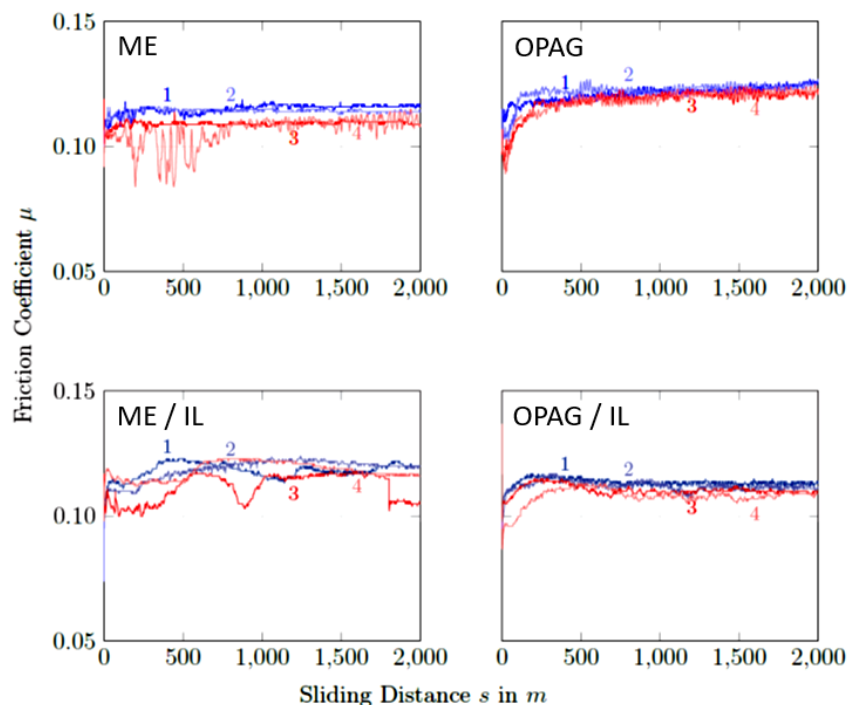


Figure S2: Coefficient of friction of oils and mixtures with IL at 90°C

4. Wear scars

A contact between the ball and a plate should produce a wear scar of a circular shape. However, the elliptical shape of the wear scars is due to the 45° inclination of the plates (see **Figure S1**), thermal expansion of the setup during the experiments and tolerances. This effect is more visible for the smaller wear scars.

5. Hertzian contact pressure

The maximum Hertzian contact pressure was calculated using Hertzian equation as below (Johnson, 1987):

$$P = 21.34 \text{ N}$$

Equivalent modulus of elasticity:

$$\frac{1}{E^*} = \frac{1-\nu_1^2}{E_1} + \frac{1-\nu_2^2}{E_2}, \quad (1)$$

$$\begin{aligned} \nu_1 &= \nu_2 = 0.3 \\ E_1 &= E_2 = 210 \text{ GPa} \end{aligned}$$

Equivalent radius:

$$\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2}, \quad (2)$$

$$\begin{aligned} R_1 &= 6.35 \text{ mm} \\ R_2 &= \infty \end{aligned}$$

Maximum Hertzian contact pressure:

$$p_{max} = \sqrt[3]{\frac{6 P E^{*2}}{\pi^3 R^2}} = 1.11 \text{ GPa} \quad (3)$$